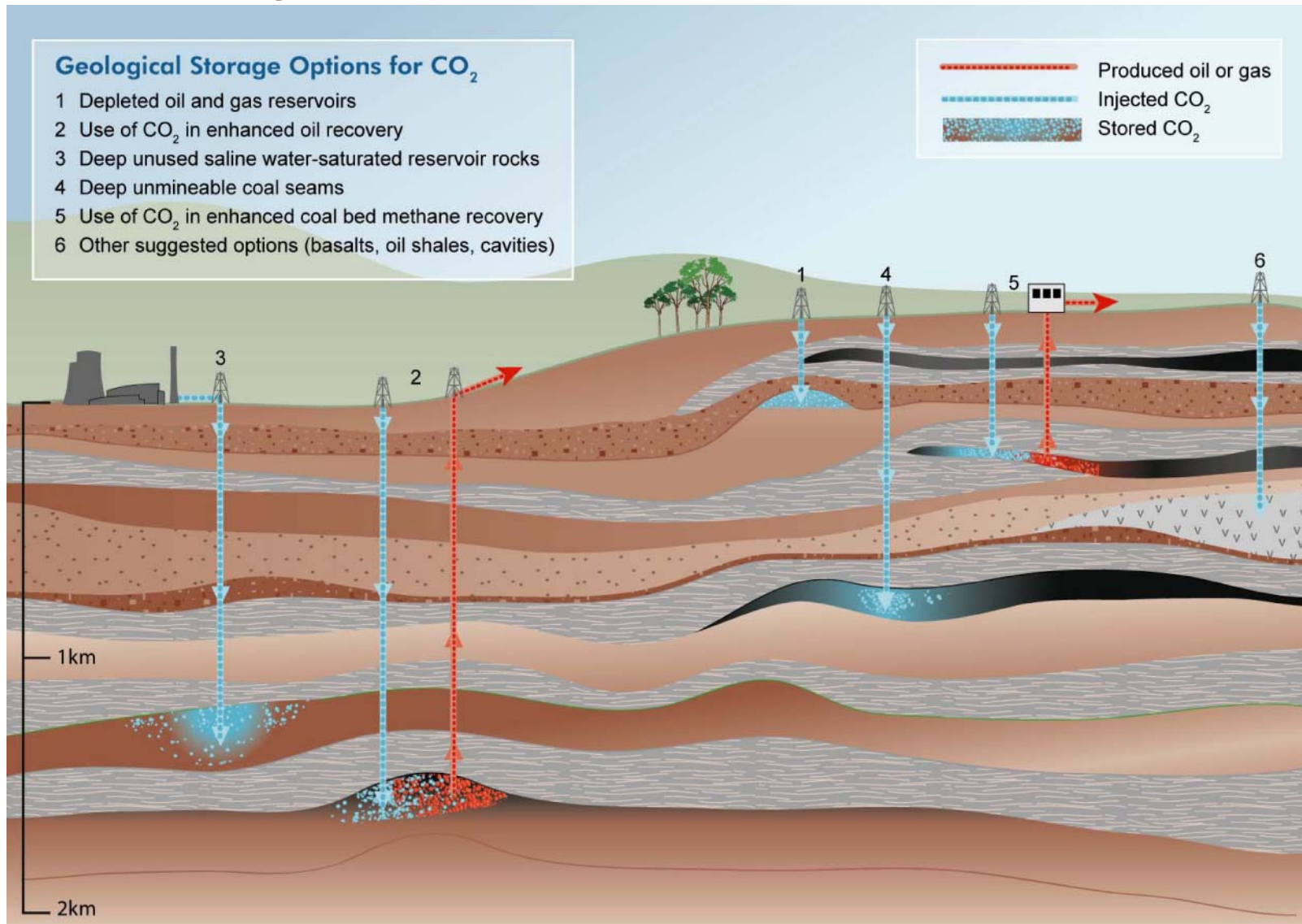


# Carbon Sequestration in Geologic Environments

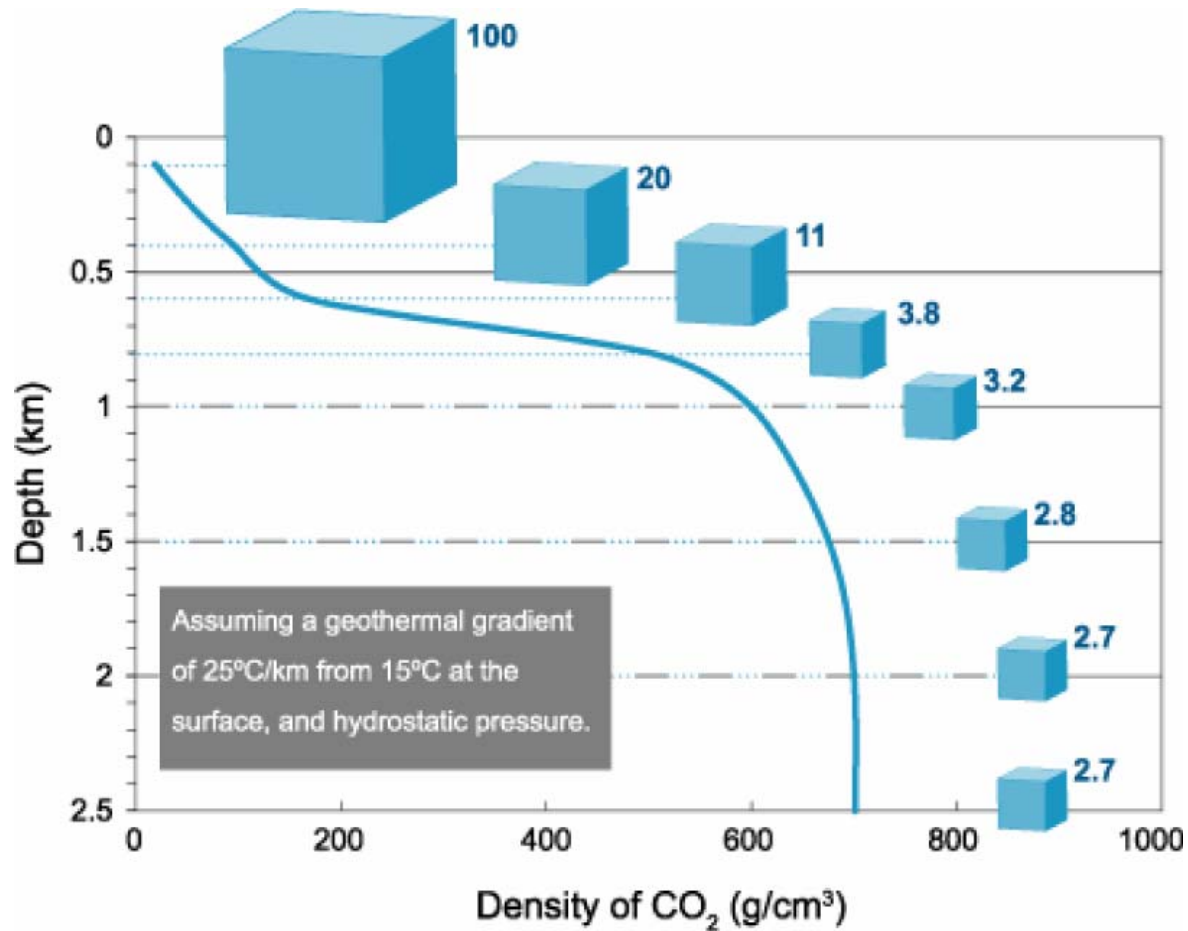
Robert M. Holt

Geology and Geological Engineering

# Geological Sequestration Options



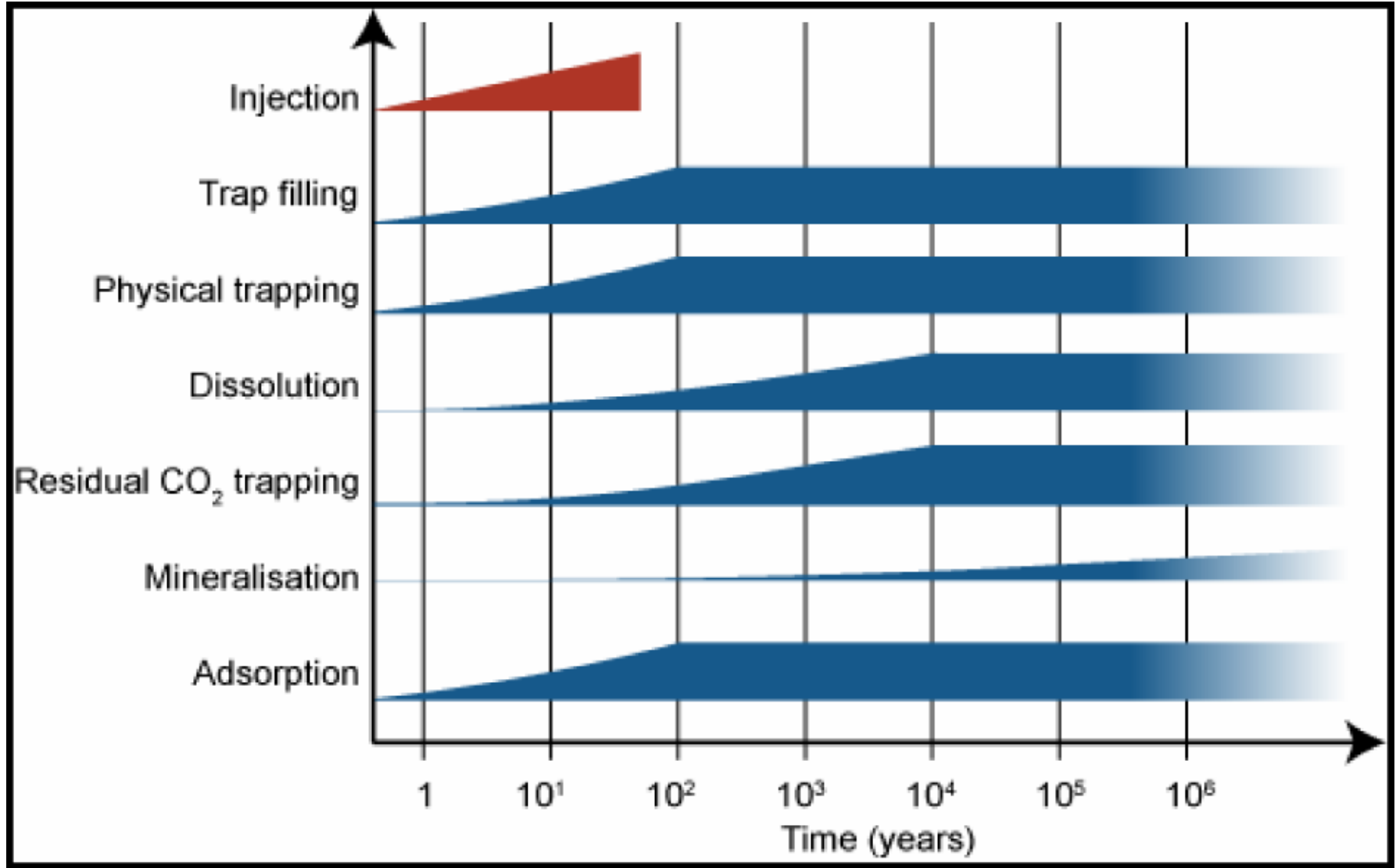
# CO<sub>2</sub> Volume Reduces With Deeper Sequestration



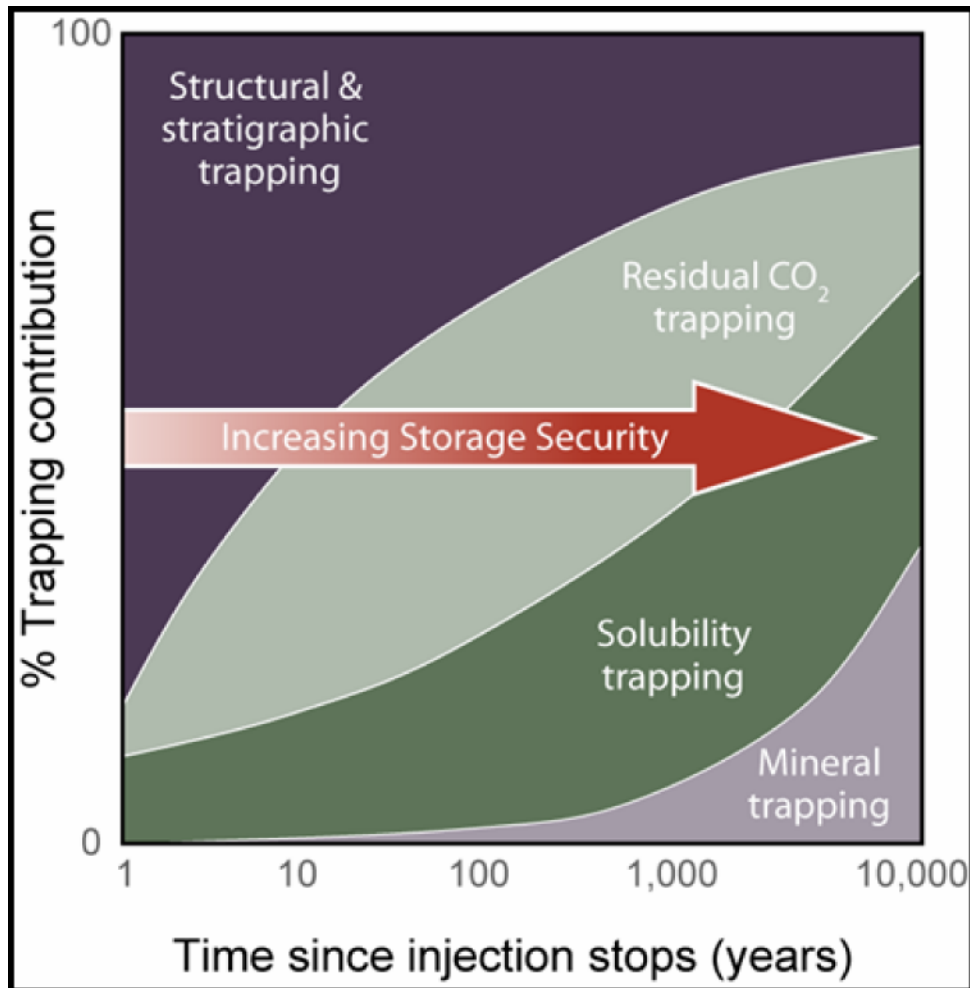
# CO<sub>2</sub> Trapping Mechanisms

- Structural and Stratigraphic Trapping
  - Capillary Barriers
  - Permeability Barrier
- Solubility Trapping
  - CO<sub>2</sub> Forms carbonic acid by mixing with in situ fluids
- Residual Saturation Trapping
  - CO<sub>2</sub> trapped in large pores
- Mineral Trapping
  - CO<sub>2</sub> binds with minerals (adsorption)
  - CO<sub>2</sub> precipitates as minerals

# Timing for CO<sub>2</sub> Storage Mechanisms



# Security of Trapped CO<sub>2</sub> Varies with Mechanism

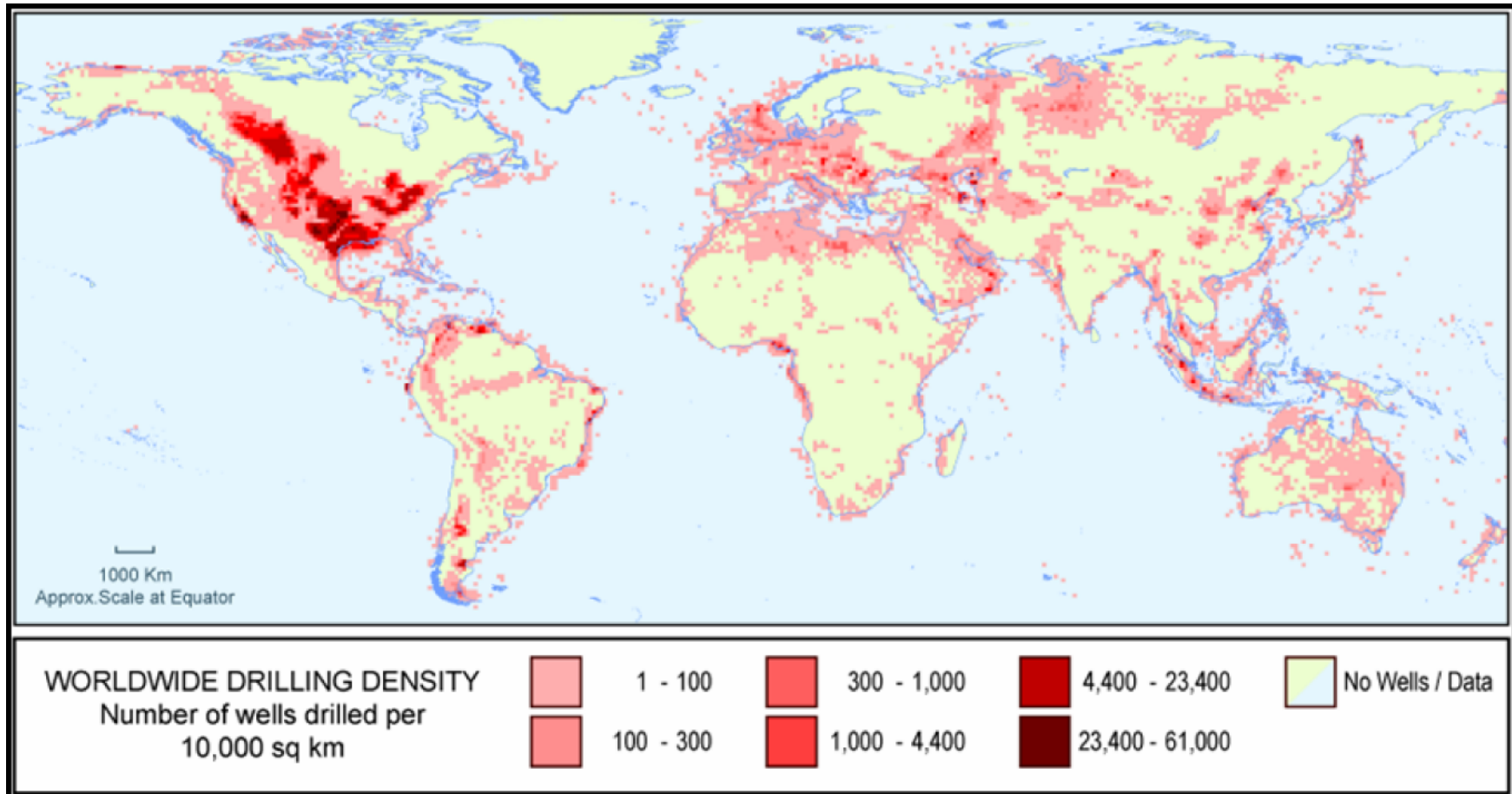


- Varies with location – different geologic and geochemical processes

# Evidence That Geologic Sequestration Could be Successful

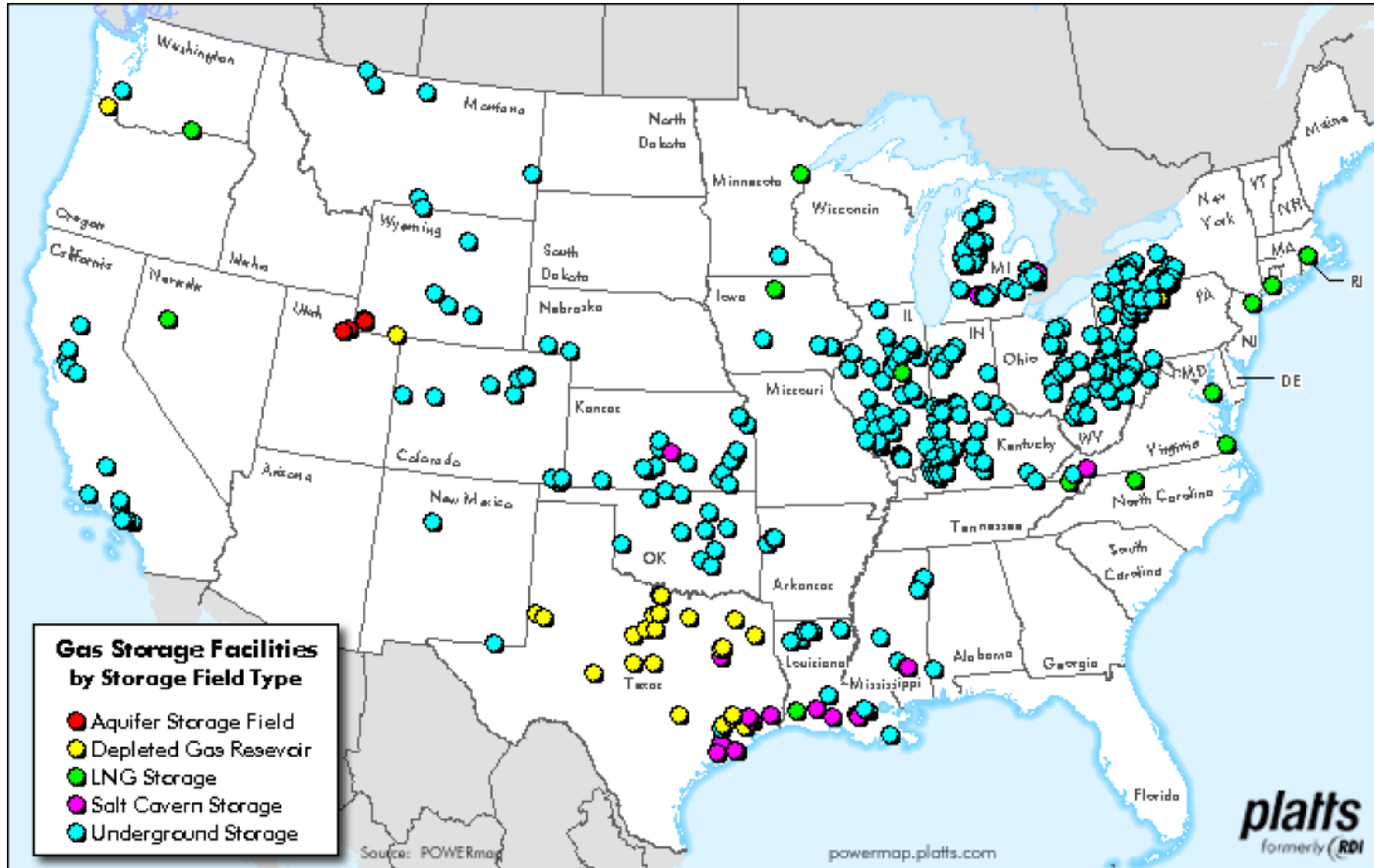
- Natural analogues
  - Oil and gas reservoirs
  - CO<sub>2</sub> formations
- Industrial analogues
  - Natural gas storage
  - Liquid waste disposal
  - CO<sub>2</sub> injection for enhanced oil and gas recovery

# World Oil and Gas Well Distribution





# Industrial Analogues

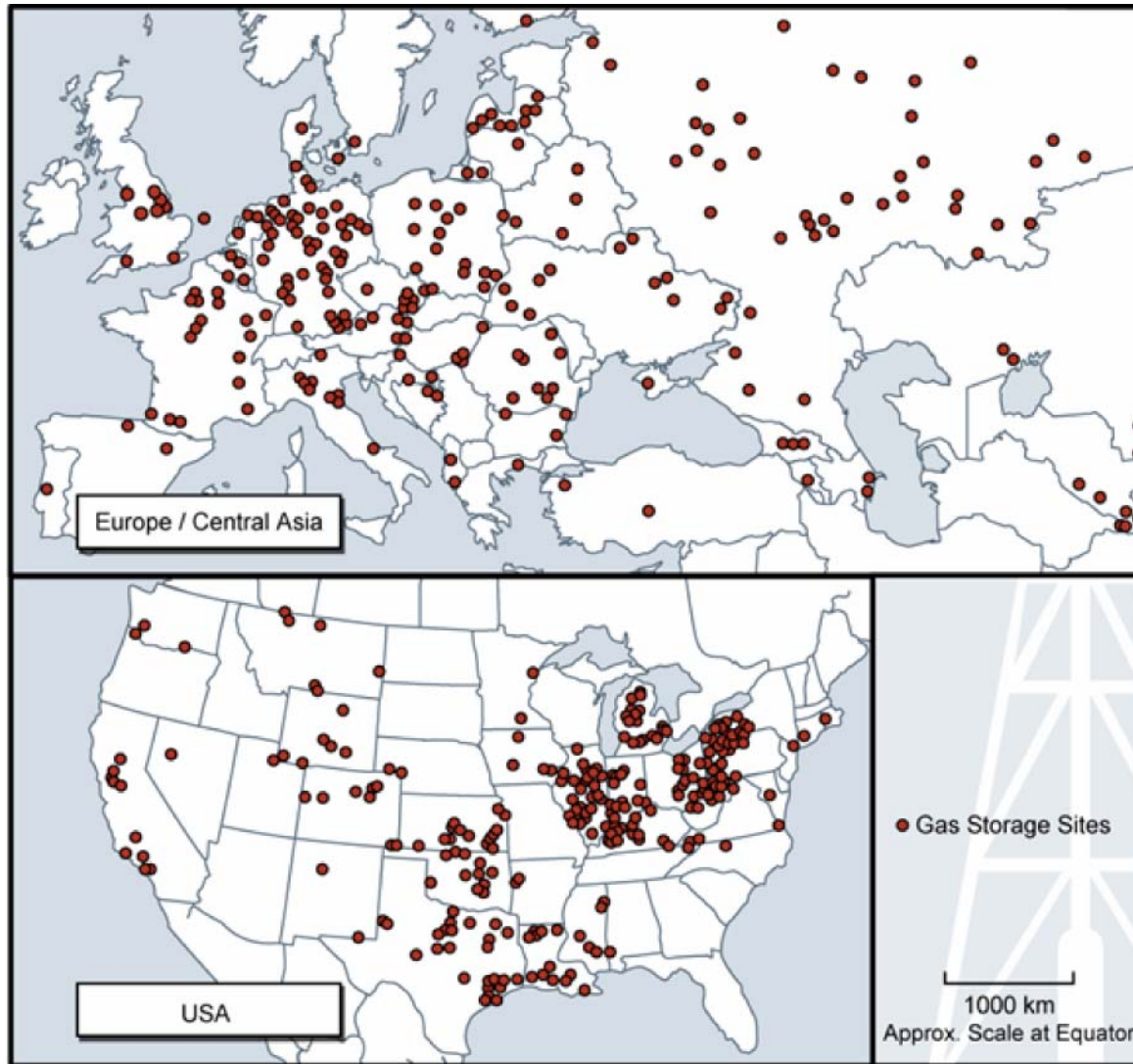


# Natural Accumulations of CO<sub>2</sub>



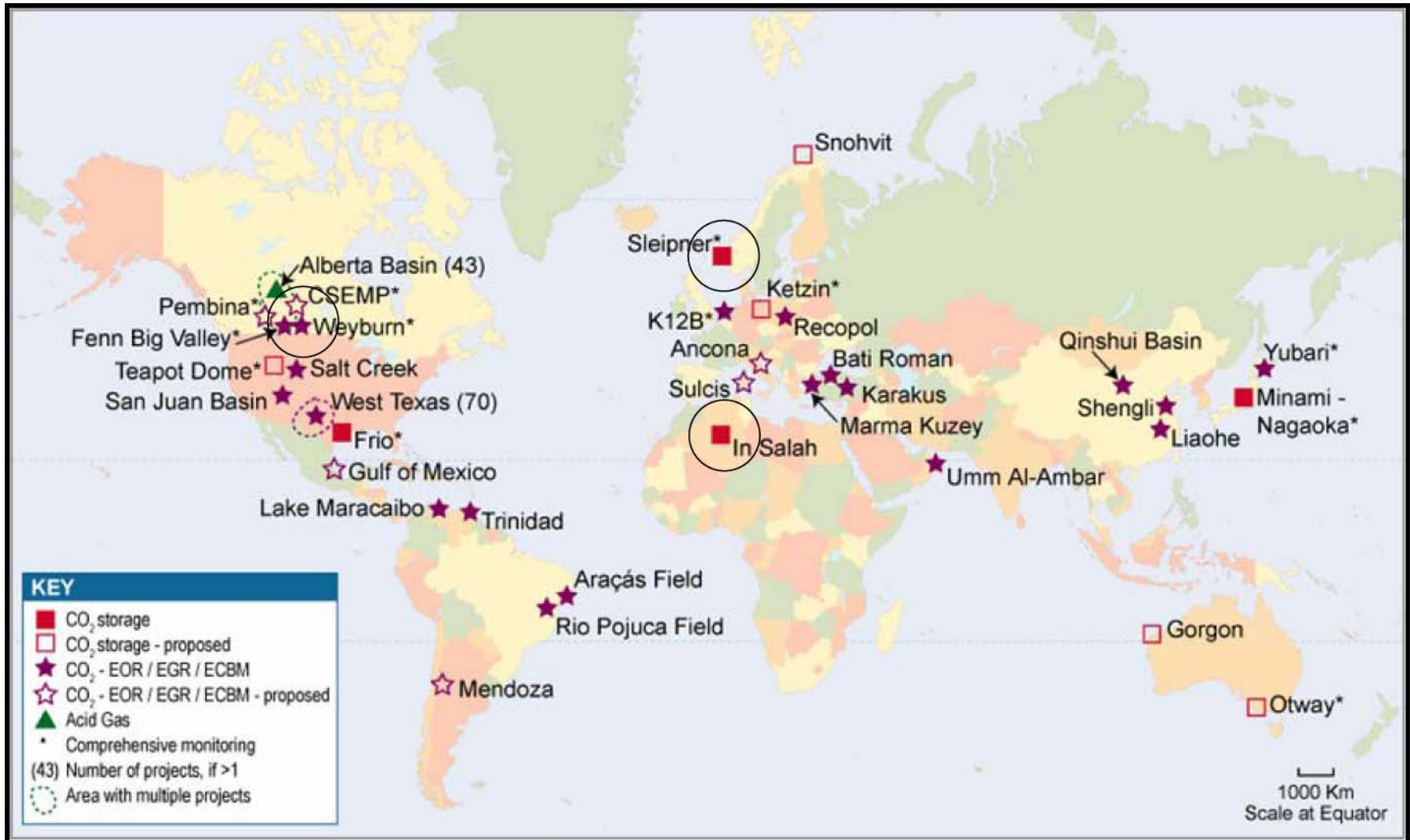
Release rates less than  $10^{-7}$ /year

# Natural Gas Storage Projects

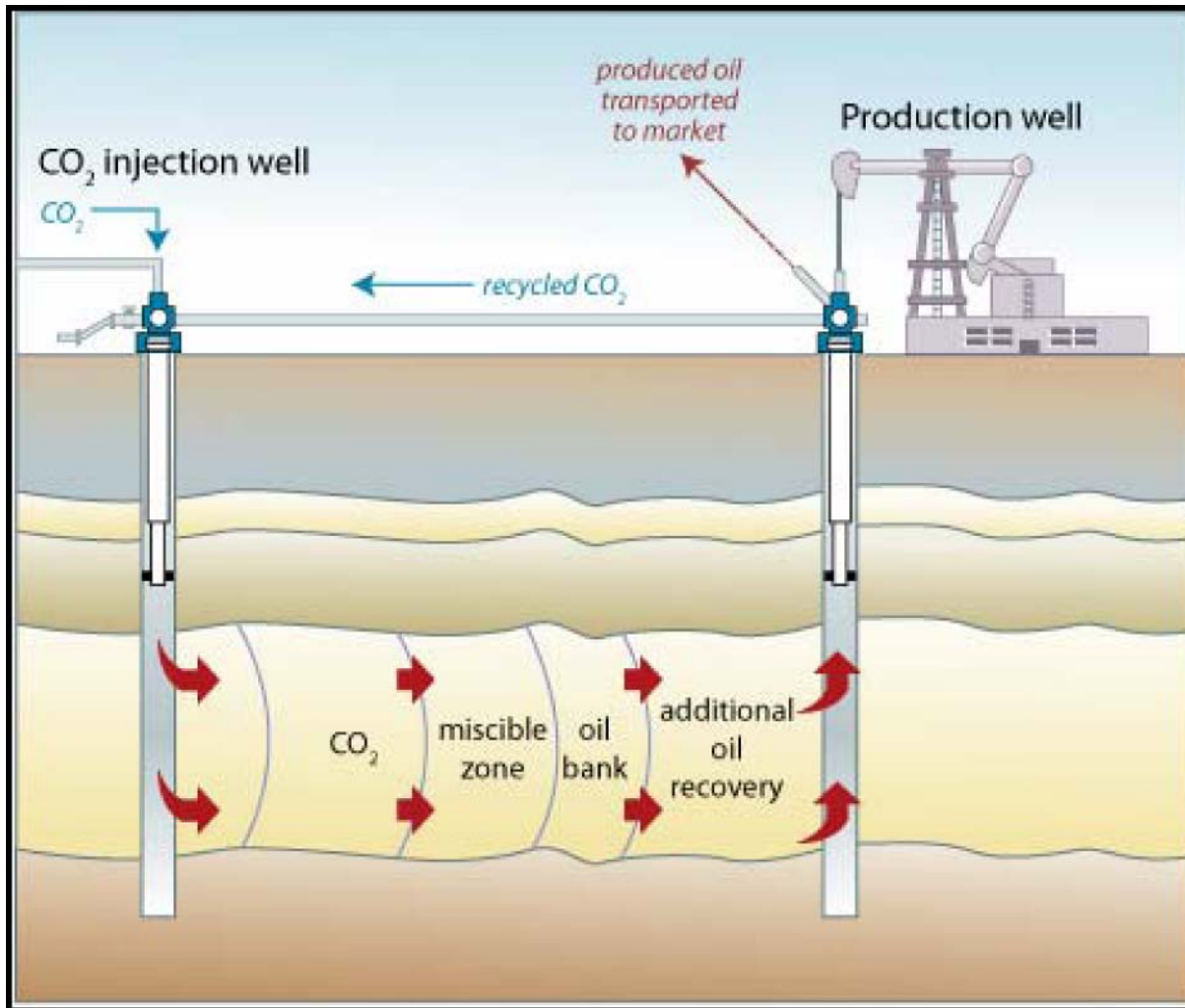


- Projects have more than 10,000 facility-years of operations
- Release rates  $<10^{-4}$  to  $10^{-6}/\text{yr}$

# Existing Gas Storage Sites

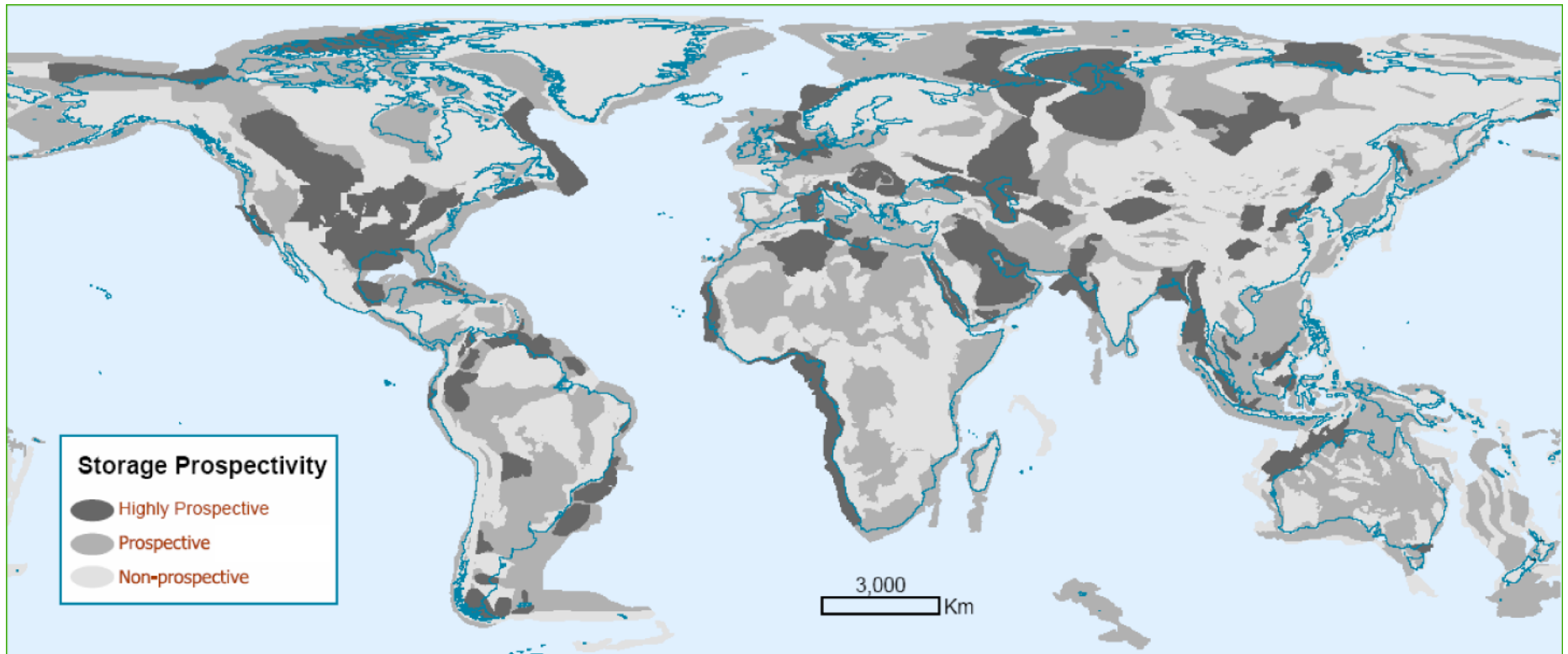


# Injection of CO<sub>2</sub> for Enhanced Recovery of Hydrocarbons



- >100 Mt injected
- Limited data show flux near zero

# World-Wide Geologic Storage Potential



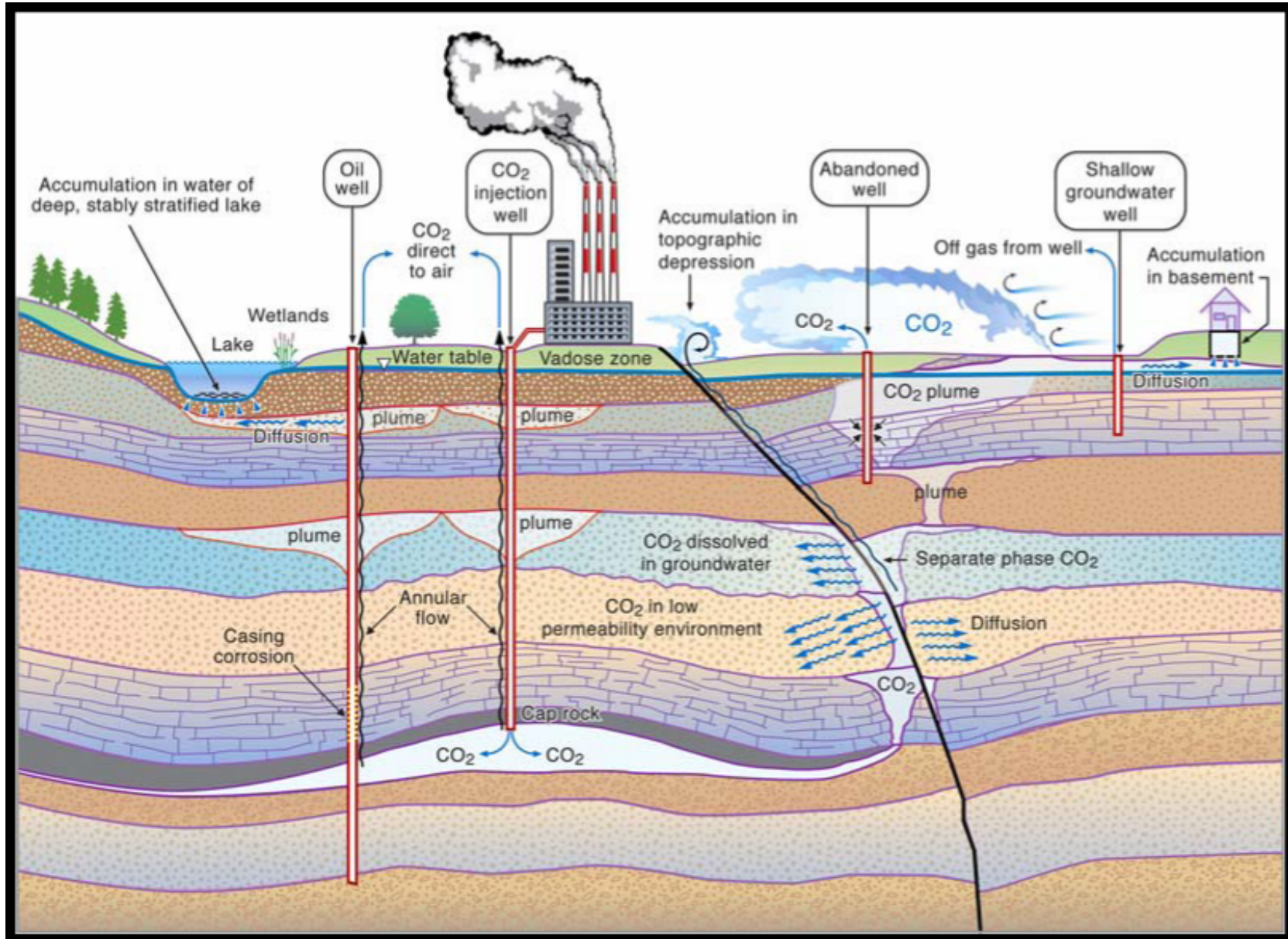
# Capacity for Geologic Sequestration

Reservoir Type	Lower Estimate of Storage Capacity (GtCO <sub>2</sub> )	Upper Estimate of Storage Capacity (GtCO <sub>2</sub> )
Oil and gas fields	675 <sup>a</sup>	900 <sup>a</sup>
Unminable coal seams (ECBM)	3–15	200
Deep saline formations	1000	Uncertain, but possibly 10 <sup>4</sup>

a. Estimates would be 25% larger if undiscovered reserves were included.

*“Available evidence suggests that worldwide, it is likely that there is a technical potential of at least about 2,000 GtCO<sub>2</sub> (545 GtC) of storage capacity in geological formations.”*

# Potential Release Pathways



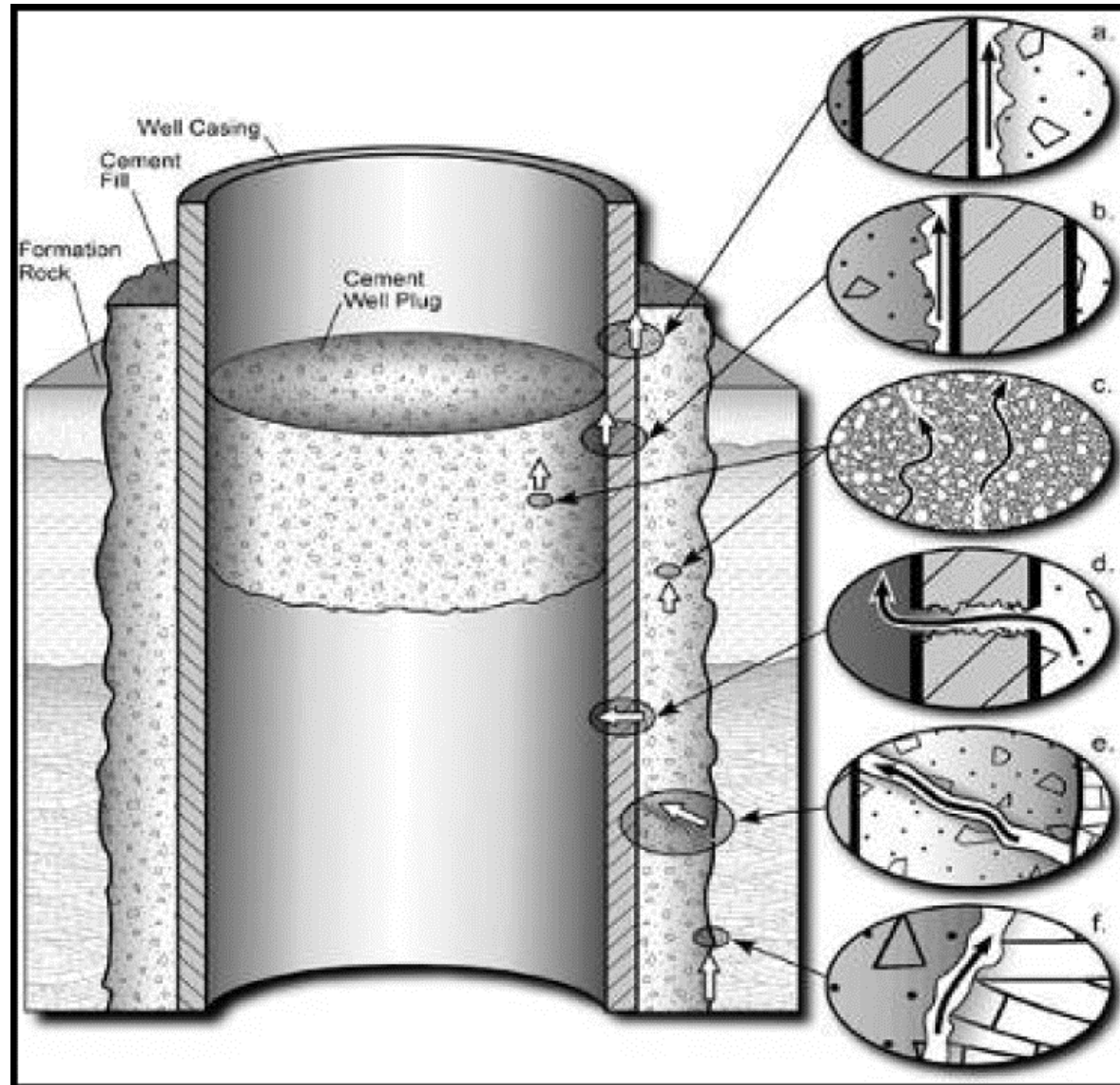


# Many Oil and Gas Fields Have Numerous Abandoned Wells



Near Andrews, Texas

# Leakage Pathways in Abandoned Wells



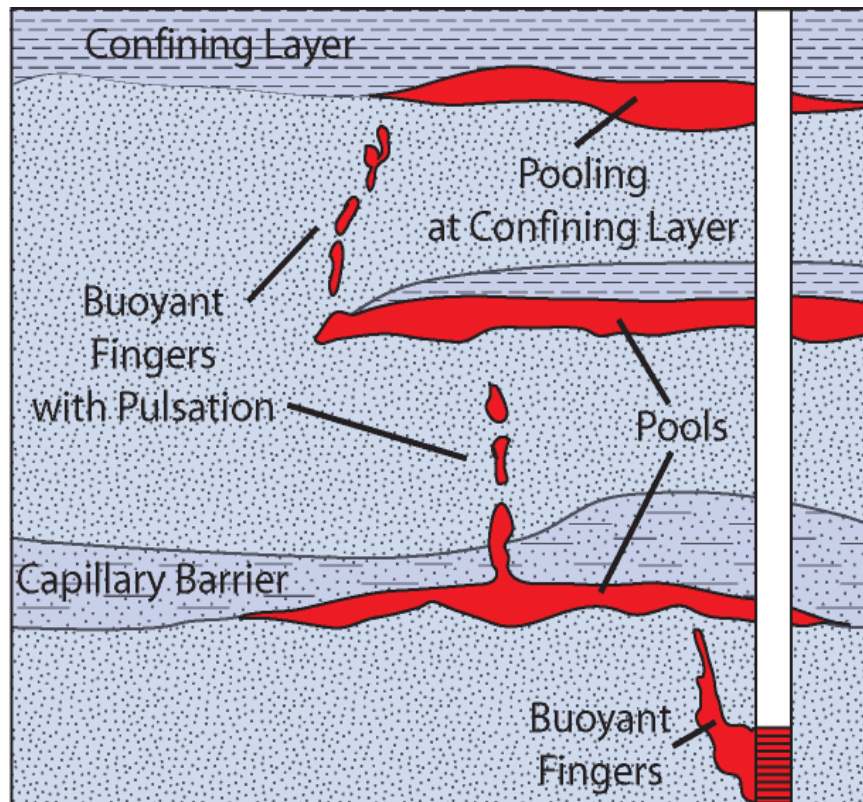
# Storage Cost Estimates

			US\$/tCO <sub>2</sub> stored		
Option type	On or offshore	Location	Low	Mid	High
Saline formation	Onshore	Australia	0.2	0.5	5.1
Saline formation	Onshore	Europe	1.9	2.8	6.2
Saline formation	Onshore	USA	0.4	0.5	4.5
Saline formation	Offshore	Australia	0.5	3.4	30.2
Saline formation	Offshore	N. Sea	4.7	7.7	12.0
Depleted oil field	Onshore	USA	0.5	1.3	4.0
Depleted gas field	Onshore	USA	0.5	2.4	12.2
Disused oil/gas field	Onshore	Europe	1.2	1.7	3.8
Disused oil/gas field	Offshore	N. Sea	3.8	6.0	8.1

**Monitoring costs - \$0.16 to \$0.30/ton**

# Development of Monitoring Technologies at the University of Mississippi

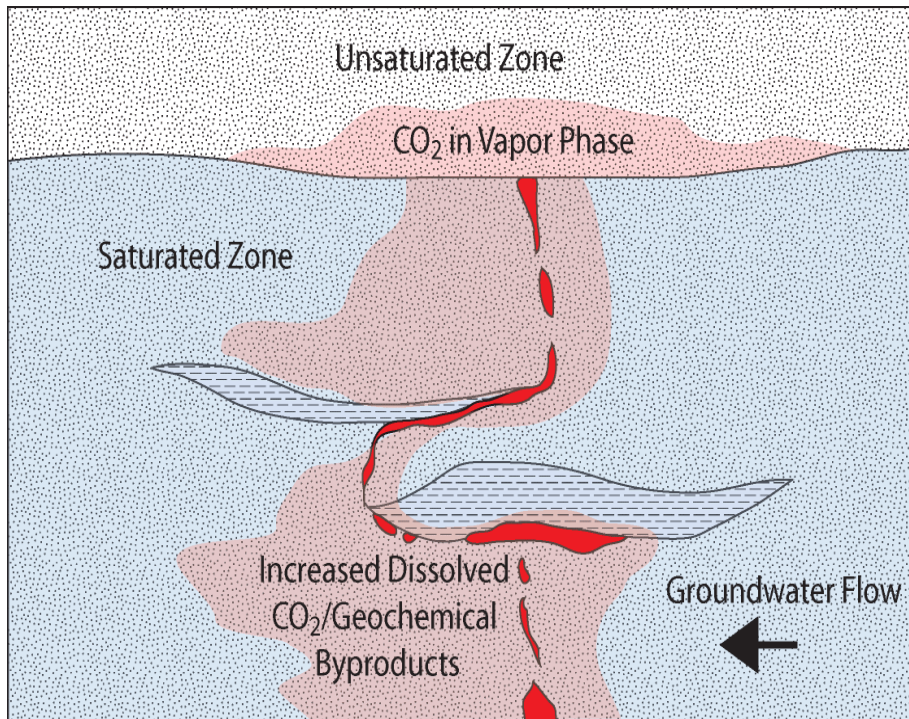
## Conceptual model for CO<sub>2</sub> transport in the saturated zone



- Buoyancy-driven fingering (with pulsation) through coarse layers
- Trapping behind fine-grained layers
- Fingering and breakthrough when CO<sub>2</sub> pressure exceeds the non-wetting phase entry pressure
- Trapping beneath aquifer confining layers with the development of large pools, or
- For phreatic aquifers, movement into the unsaturated zone and pooling above the capillary fringe

# Monitoring Requires Observations of CO<sub>2</sub> and its Geochemical Byproducts

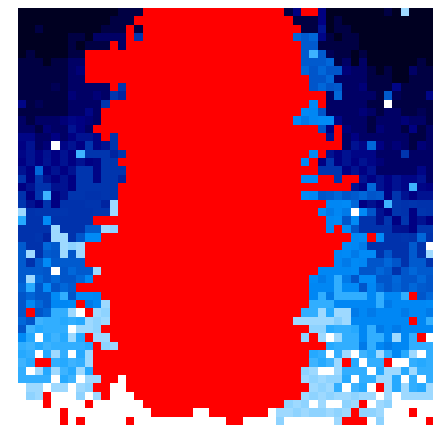
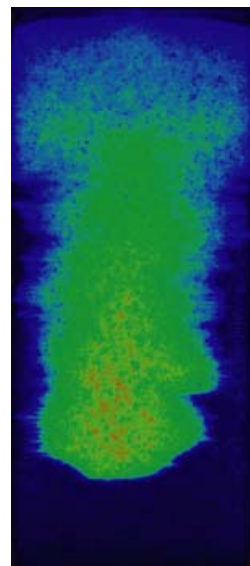
## *Basic science questions:*



- 1) What controls CO<sub>2</sub> partitioning and dissolution into the aqueous phase? (*surface area of CO<sub>2</sub> blobs*)
- 2) What chemical reactions will occur and what are their rates? (*grain surface chemistry, reaction rates*)
- 3) What is the size of zones of detectable CO<sub>2</sub> and byproducts? (*capillary heterogeneity, dissolution rates, reaction rates, ambient flow, source term*)
- 4) What type of monitoring design will be required to insure detection? (*source term size, heterogeneity, reaction/dissolution rates, action level*)

# Bench-Scale Experiments to Answer Basic Science Questions 1-3 and Improve Field-Scale Tracer Test Design

- Use transmitted light techniques to visualize CO<sub>2</sub> phase structure evolution in heterogeneous sand chambers
- Quantify CO<sub>2</sub> dissolution into the aqueous phase and monitor geochemical evolution of by products in aqueous phase
- Simulate experimental results using continuum and Macroscopic Invasion Percolation modeling approaches
- Select modeling approach for field-scale tracer test design



# Field-Scale Tracer Tests to Answer Basic Science Questions 3-4

- Define background chemistry of groundwaters and CO<sub>2</sub> content in the unsaturated zone vapor phase
- Characterize site-specific heterogeneity in the saturated and unsaturated zones
- Use simulations to design tracer tests and determine saturated and unsaturated zone monitoring locations
- Develop sampling procedures and protocols
- Conduct tracer tests

